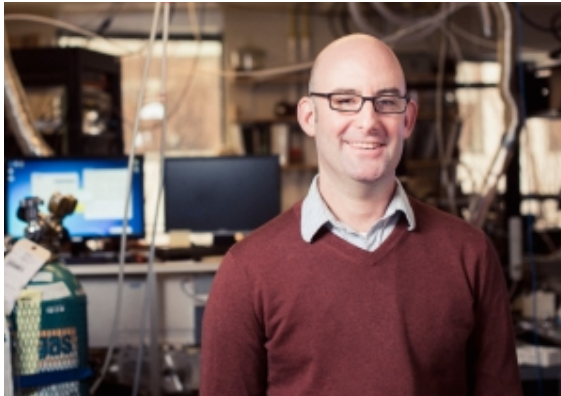




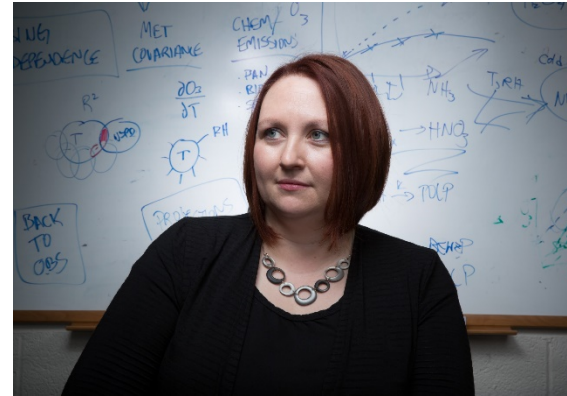
The Hawai'i Island Vog Network: Building a community-based sensor network to track vog emissions



Introductions: MIT Team



Jesse Kroll
(Principal Investigator)



Colette Heald
(co-Principal Investigator)

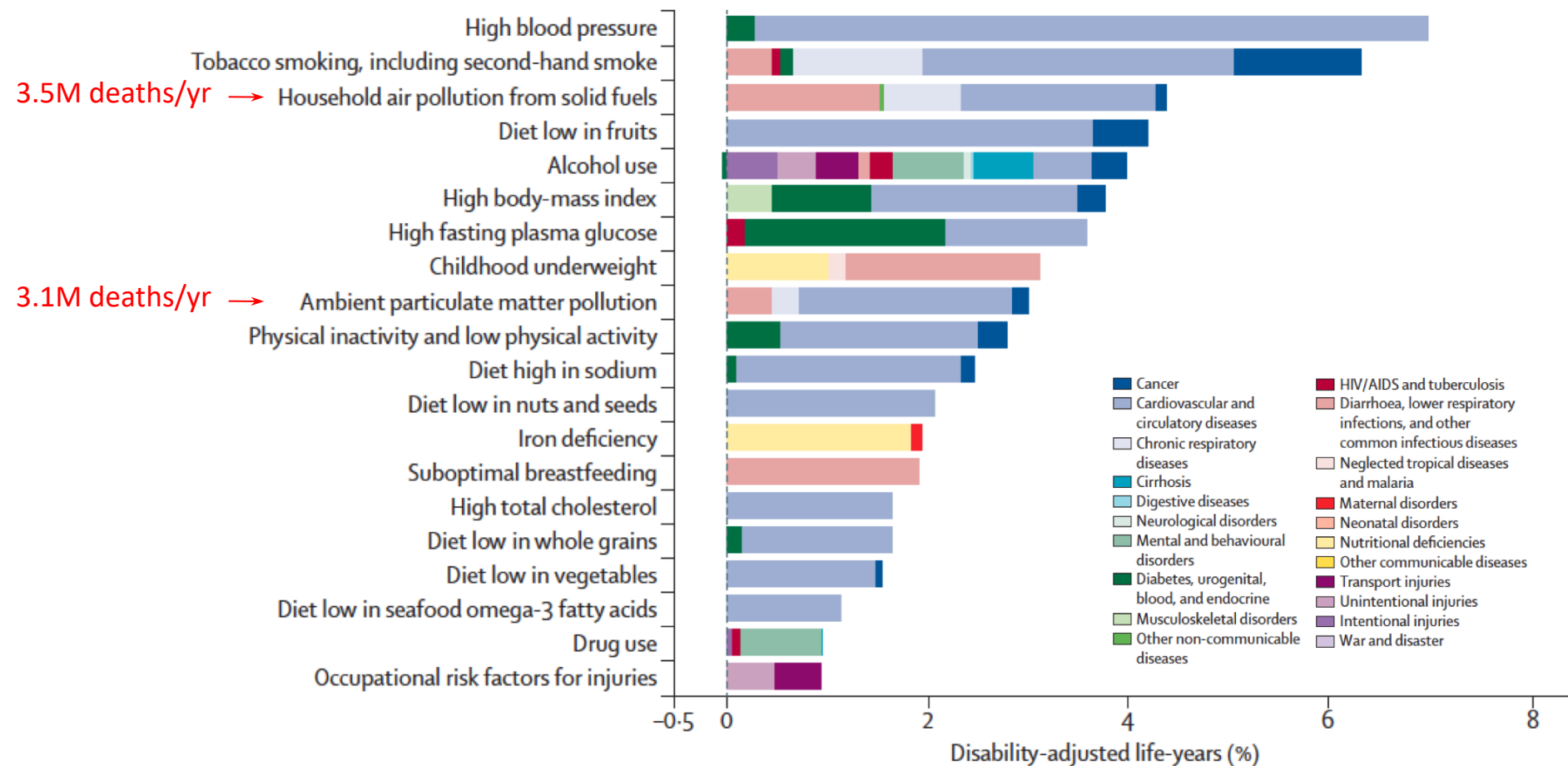


Ben Crawford
(Postdoc, Local Project Lead)



Kathy Vandiver
(Education Coordinator)

Aerosol Pollution is a Significant Public Health Concern



Project Origins: MIT class

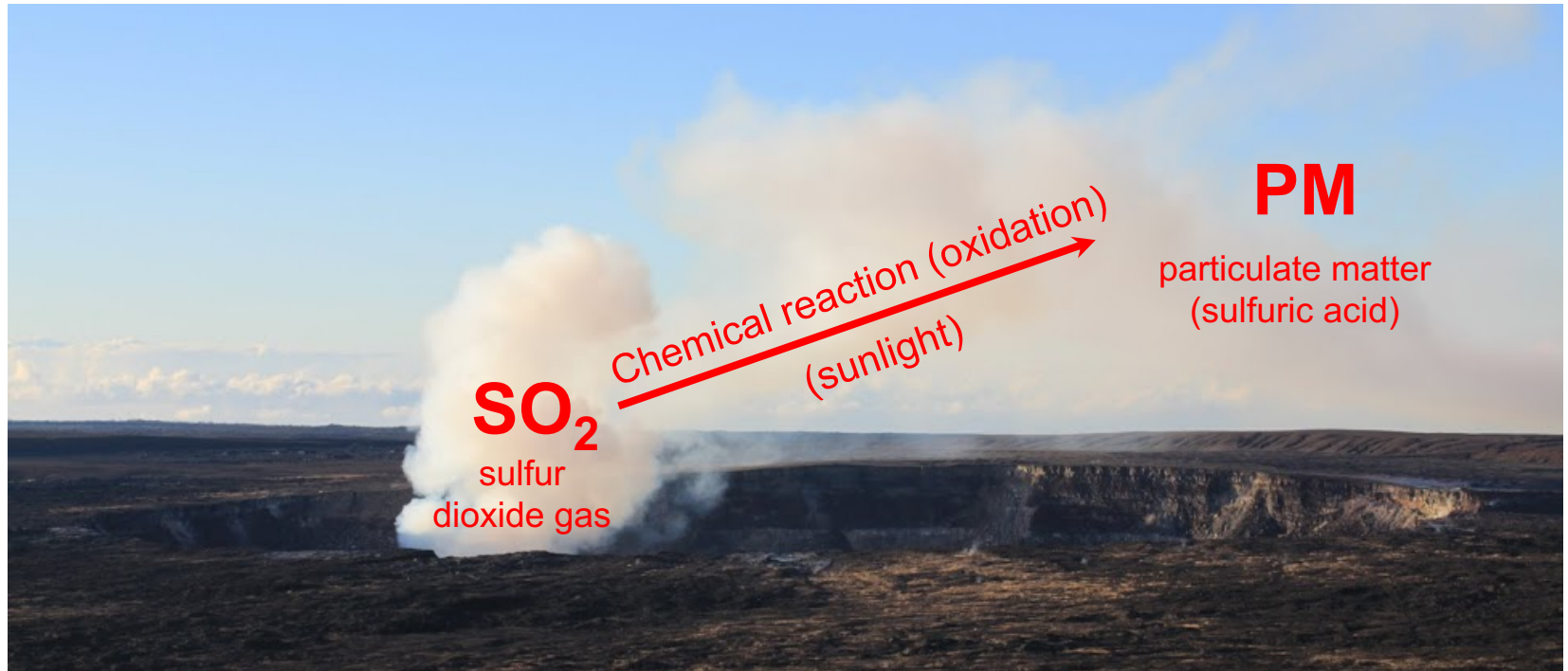
“Traveling Research Environmental Experiences” (TREX)

January-term undergraduate class in MIT's
Department of Civil and Environmental Engineering



research in Hawai'i since 2001
studies of vog since 2012 (77 students)

Volcanic emissions (“vog”)



~4000 tons SO₂/day
(largest single source in the US)
toxic to humans, plants

unhealthy to breathe
highly acidic

Health impacts of vog

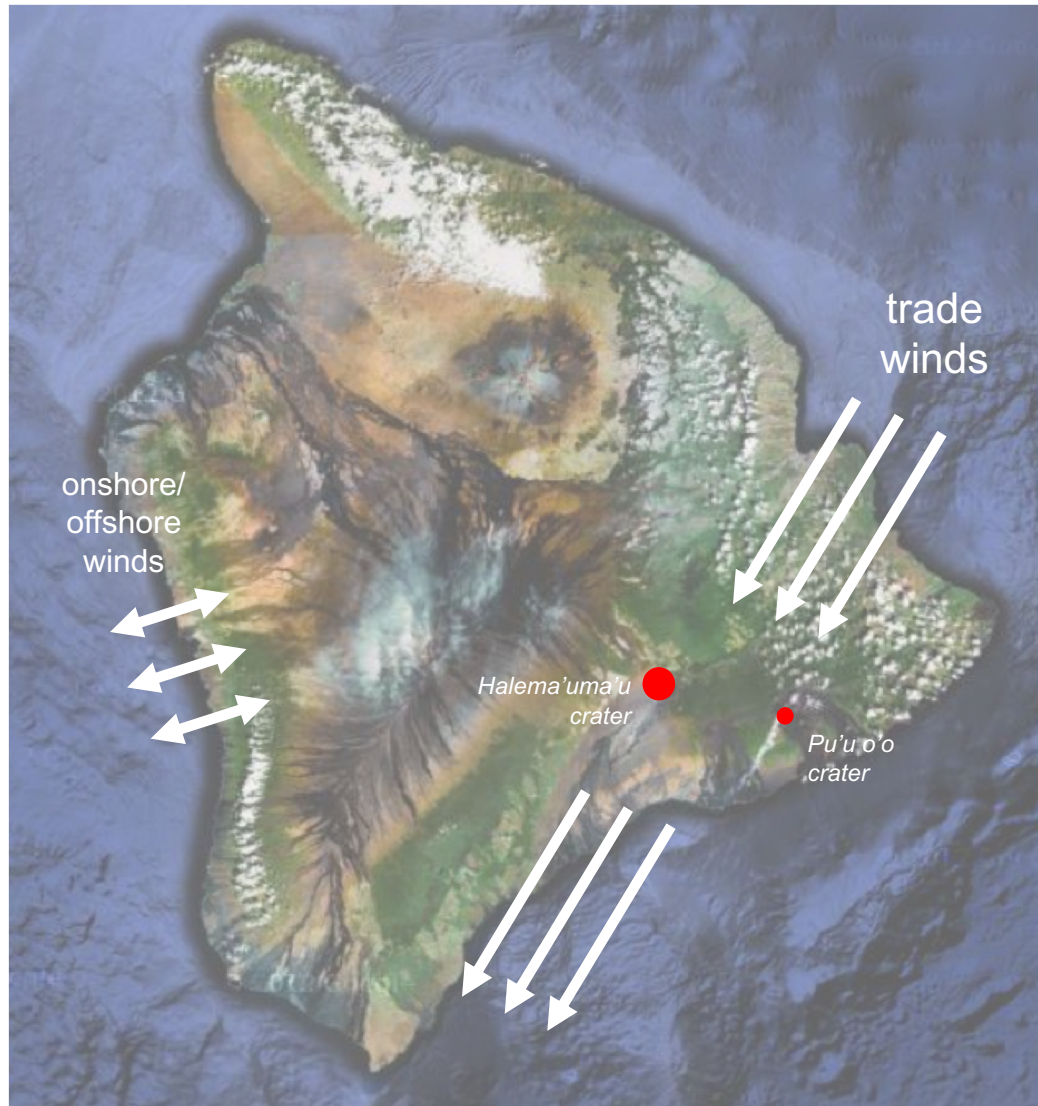
Vog exposure is correlated with:

- increased blood pressure
- diagnosed hypertension
- increases in outpatient clinic visits for cough, headache, pharyngitis, and acute airway problems
- self-reported cough, phlegm, rhinorrhea, sore/dry throat, shortness of breath, sinus congestion, continual wheezing, eye irritation, skin irritation...

Among children, exposure to acidic PM is correlated with:

- cough, decreased FEV1/FVC (decreased pulmonary function)
- but not asthma or bronchitis

Air quality in Hawai'i



Major pollutant of concern depends on location:

Hilo side - “fresh” vog - **SO₂**

Kona side - “aged” vog - **PM**

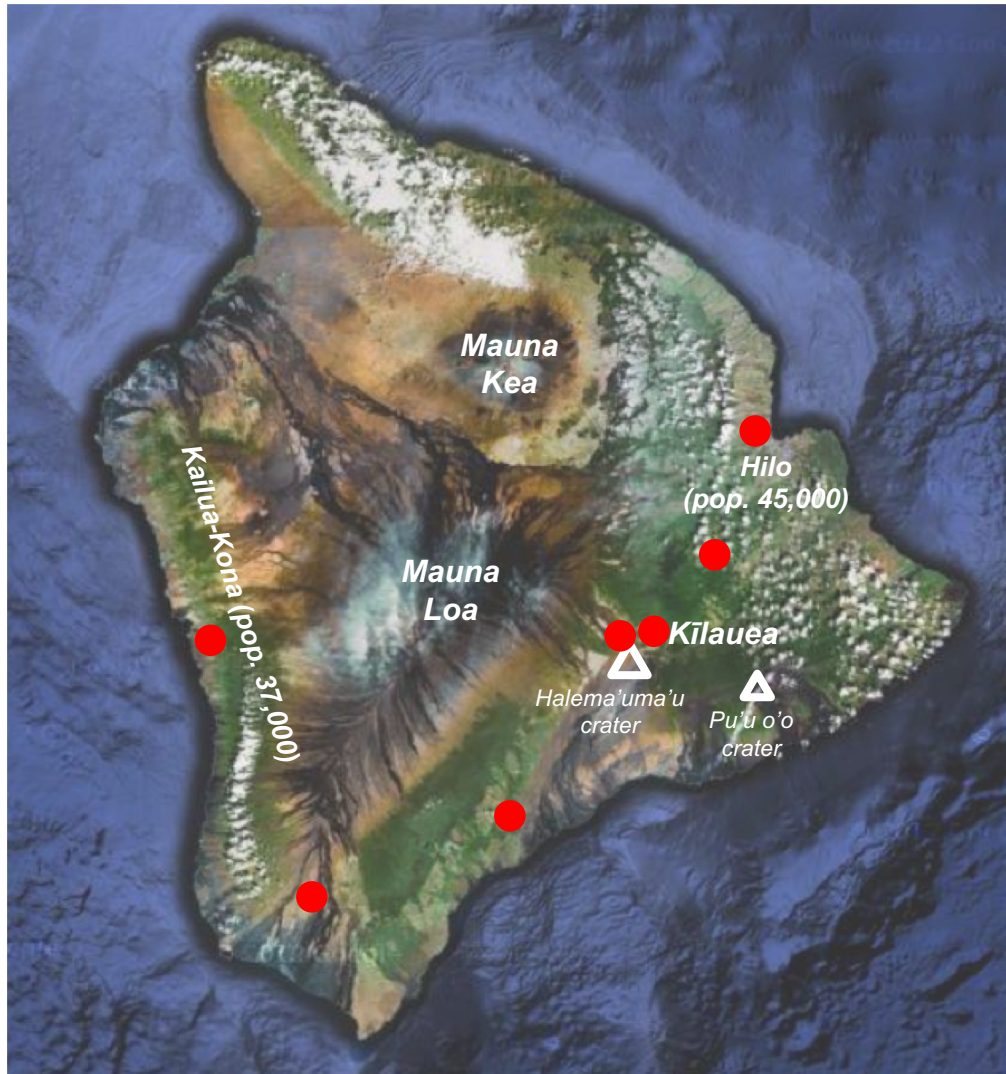
How to measure air quality?

- Research/regulatory grade instrumentation
- **Advantage:** precision and accuracy
- **Challenge:** expense (~\$10K per sensor)
 - Limited spatial coverage -> where to place instruments?
 - Representative of a large area
 - Different purposes (e.g. research vs public health)



Met One Instruments, Beta Attenuation Monitor (BAM)

Air quality monitoring in Hawai'i



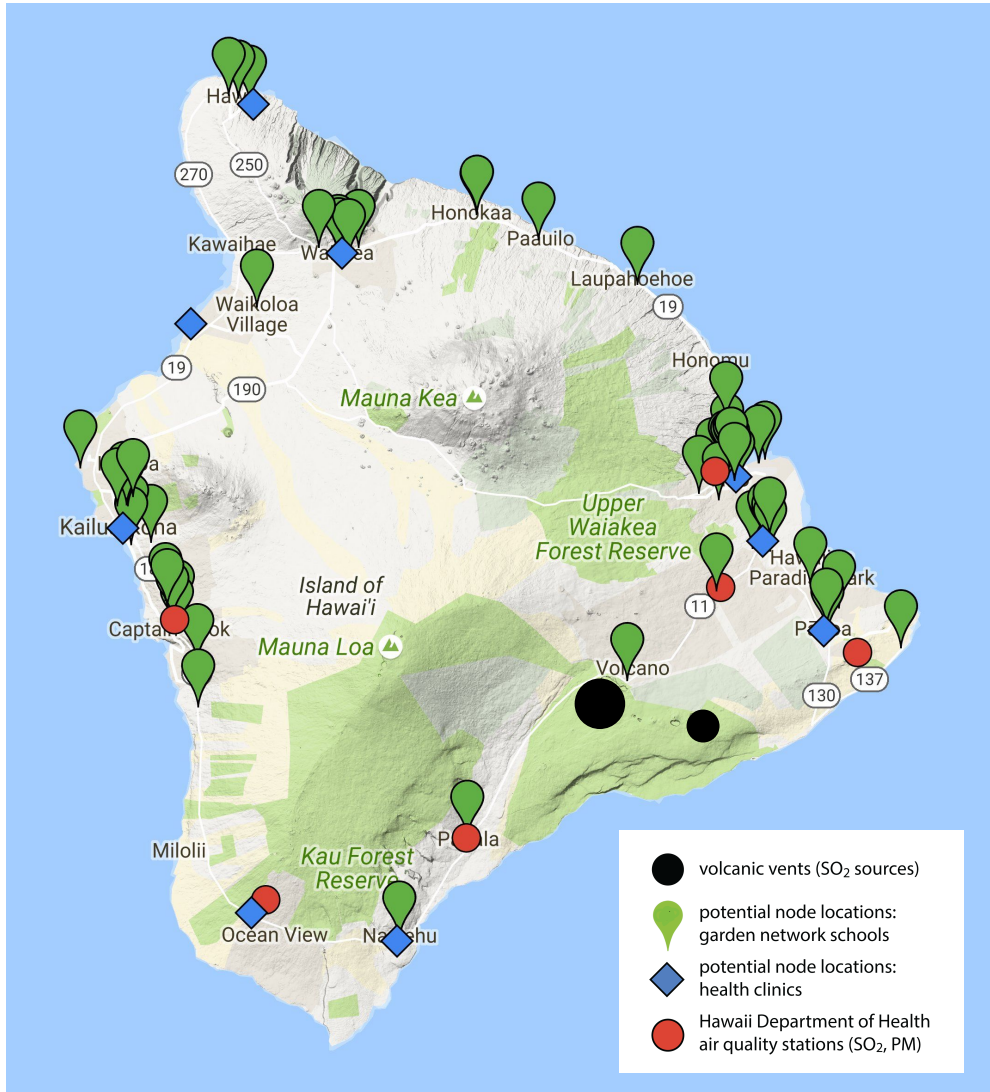
7 SO₂/PM monitoring sites
(5 Hawai'i Department of
Health, 2 NPS/USGS)

Low-cost approach

- **Advantage:** inexpensive (~\$200 per sensor)
 - Dense spatial coverage
- **Challenge:** sensor performance needs to be studied
 - Calibration & maintenance for a large number of sensors



Planned low-cost network



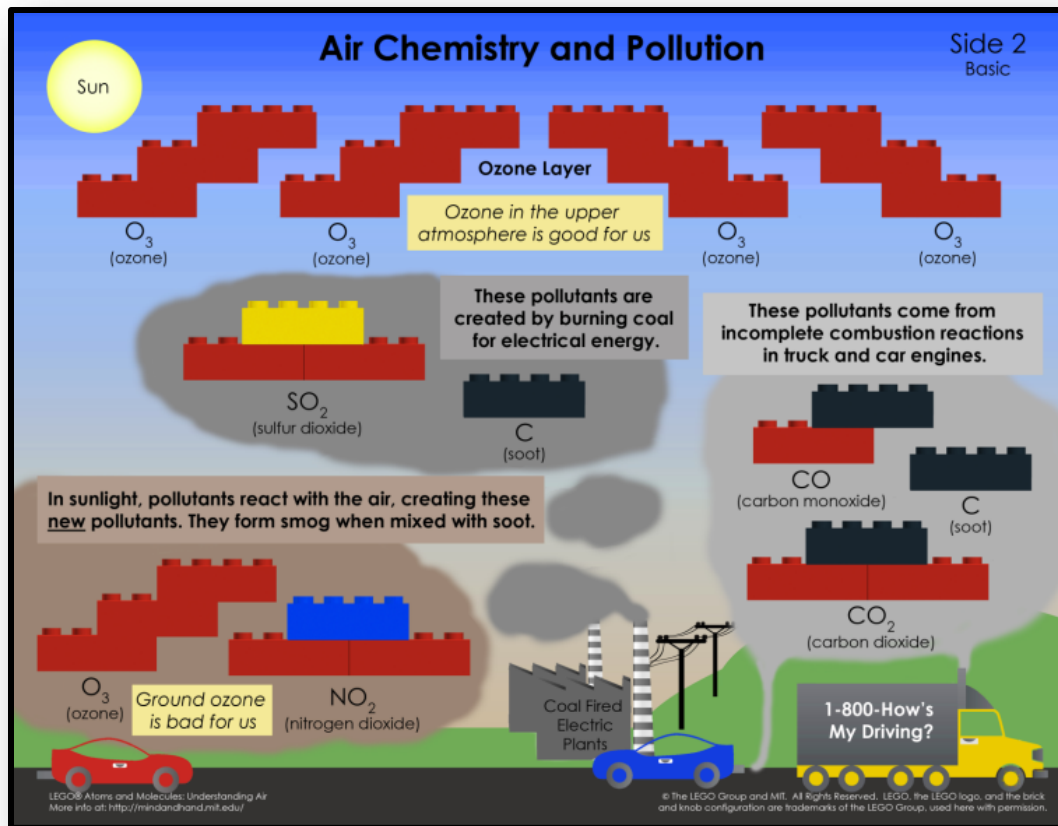
- Several dozen sensor nodes for measuring V_{og} components and meteorological parameters
- Primarily located at schools (green pins), local health clinics (blue diamonds)
- Expected deployment: fall 2018
- Data sent in near-real-time to a server for online access via a public web portal
- **Intended as a public resource**

Education and Engagement



- January 2018: Project team (Kroll, Heald, Vandiver, Crawford) traveled to Hawai'i to discuss the project with teachers, health professionals, and community members
 - 3 teacher workshops (~50 educators total)
 - 4 meetings at health clinics (~10 health professionals)
 - public talk, several meetings with individuals
- Received offers to “host” sensor nodes, useful feedback on how the project can relate to educational curricula

Education and Engagement : MIT Edgerton Center Molecule Set



A hands-on STEM package for Teaching Key Concepts in Biology, Chemistry and Earth Science










*Included in the PD Course planned for this summer
With the Air Quality Sensor Network!*

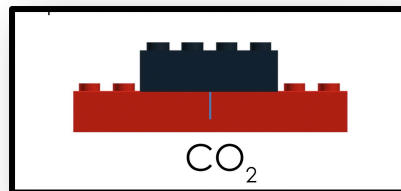
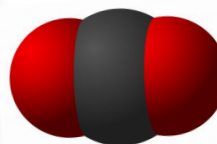
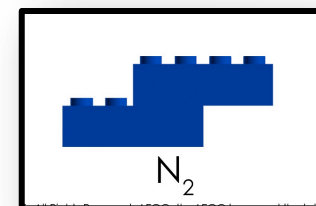
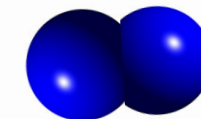
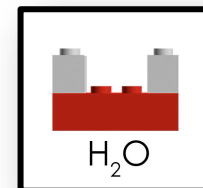
This curriculum is designed to introduce students to the Atomic Nature of Matter

With examples from

- Chemistry
- Biology
- Earth Sciences

A LEGO® brick represents an atom.
The key uses CPK chemistry colors.

Hydrogen (H)	=	
Sodium (Na)	=	
Calcium (Ca) or Magnesium (Mg)	=	
Iron (Fe) or Copper (Cu)	=	
Carbon (C)	=	
Nitrogen (N)	=	
Sulfur (S)	=	
Oxygen (O)	=	
Chlorine (Cl)	=	



Chemical reactions demonstrated in three subject areas:



Chemistry: Chemical Reactions

Model the chemical reaction just experienced as a wet lab:

baking soda + calcium chloride produce sodium chloride, carbon dioxide, water and chalk.



Biology: Photosynthesis

Model the reaction for photosynthesis: water and carbon dioxide produce sugar and oxygen. (The reaction for cellular respiration – the reverse can be modeled too!)



Earth Science: Understanding Air

Model air as a mixture of gases.

→ Model combustion using hydrocarbons + oxygen to produce carbon dioxide + water.

Pollutants such as CO and C appear if less oxygen is available.

Lego chemical reaction example

Years ago, CO₂ in the air measured 350 ppm (parts per million).


In January 2018, CO₂ in the air measured 406 ppm!

See how burning fuel continues to create more CO₂ below?

**Reaction
Occurs!**

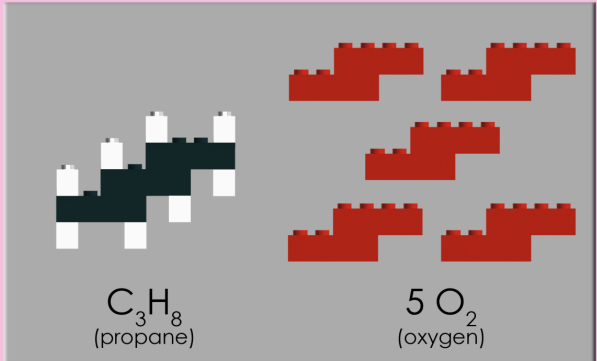
1) Build propane and oxygen.
Place the atoms on their pictures below.

2) Rearrange the same atoms
into these products. Place them
on their pictures below.

 **Burning Fuel**
Complete Combustion

Side 1
Reactants

Combustion is a chemical reaction.
Build the fuel and oxygen molecules with LEGO bricks. Place them on their pictures.



C_3H_8
(propane)

$5 O_2$
(oxygen)

spark
(TURN OVER)

LEGO® Atoms and Molecules: Understanding Air
More info at: <http://mindandhand.mit.edu/>

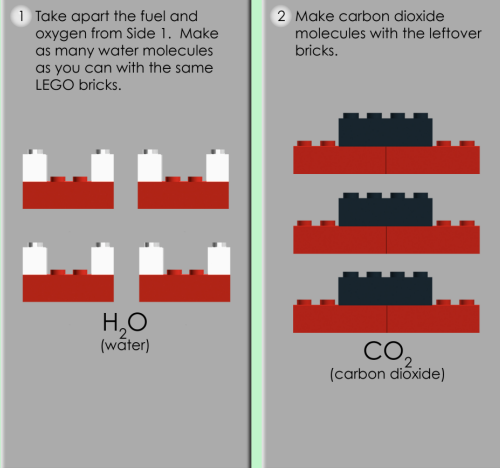
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Burning Fuel
Complete Combustion

Side 2
Products

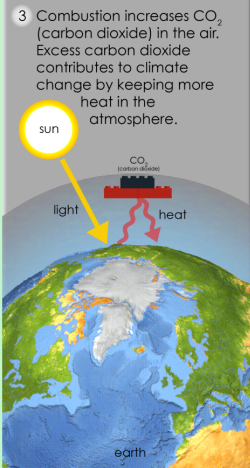
When there is plenty of oxygen available, fuel burns completely, producing only carbon dioxide and water.
This reaction is called **complete combustion**.

- 1 Take apart the fuel and oxygen from Side 1. Make as many water molecules as you can with the same LEGO bricks.
- 2 Make carbon dioxide molecules with the leftover bricks.
- 3 Combustion increases CO₂ (carbon dioxide) in the air. Excess carbon dioxide contributes to climate change by keeping more heat in the atmosphere.



H_2O
(water)

CO_2
(carbon dioxide)



LEGO® Atoms and Molecules: Understanding Air
More info at: <http://mindandhand.mit.edu/>

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Next steps / feedback

September 2018: Educator PD course (Using the sensor node & data; Atoms and Molecules)

September 2018: Ben moves to Hawai'i; Installation of sensor nodes

Please contact us if you are interested in “hosting” a sensor,
and/or if you have any ideas/thoughts/questions!

Ben: bencrawf@mit.edu

Kathy: kathymv@mit.edu

Colette: heald@mit.edu

Jesse: jhkroll@mit.edu

Thanks!!